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Research Article

Synergistic Antioxidant Activity of Clove Oleoresin with Capsicum Oleoresin and Kalonji Seeds Extract in Sunflower Oil - @

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ABSTRACT

The essential fatty acids are one of the vital nutrients to be included in the diet. However, they are susceptible to autoxidation and maintaining their stability during storage is a foremost problem. Synthetic antioxidants used at present are reported to be toxic in nature. Hence, there is a demand from consumers for foods containing natural antioxidants. Various natural products like clove oleoresin, capsicum oleoresin and kalonji seeds are known to show substantial antioxidant activity. Their activity can be improved by studying structural aspects and evaluating their synergism. The current investigation aims to fractionate clove oleoresin by means of column chromatography isolating the eugenol rich fraction. Though clove oleoresin showed pro-oxidant effect, its eugenol rich fraction showed marginal antioxidant activity in sunflower oil. The clove oleoresin and capsicum oleoresin, both showed good antioxidant activity in synergism with kalonji seeds extract. The study indicated the synergistic potential of natural antioxidants to be used as a substitute for synthetic antioxidants.

Keywords: Sunflower oil; Natural antioxidants; Clove; Synergistic activity; Kalonji; Capsicum

ABBREVIATIONS

EFAs: Essential Fatty Acids; TBHQ: Tertiary Butylhydroquinone; BHT: Butylhydroxytoluene; BHA: Butylhydroxyanisole; RSFO: Refined Sunflower Oil; KEE: Kalonji Seeds Ethanol Extract; ERF: Eugenol Rich Fraction; OXF: Oxidative Factor

INTRODUCTION

Vegetable oils such as flaxseed, soybean, safflower and sunflower are the richest sources of essential fatty acids (EFAs). Owing to the role of these EFAs to perform significant functions in human body and the fact that they cannot be produced in human body, EFAs need to be provided through diet [1]. Concurrently, being unsaturated, they are vulnerable to autoxidation. Ultimately, they have a very short shelf life if they are not stored with an antioxidant. The synthetic antioxidants used for the same purpose are Tertiary Butylhydroquinone (TBHQ), Butylhydroxytoluene (BHT) and Butylhydroxyanisole (BHA). Because of the unsafe nature of these synthetic antioxidants and the consumers' approach towards the natural ingredients, research is now oriented towards the search for novel natural antioxidants [2].

Several studies have analyzed the antioxidant potential of a wide variety of vegetables and fruits [3]. Extracts from spices, herbs and hulls are examined to have various degrees of antioxidant activity [4,5]. The presence of active components in spices has been demonstrated over the last 30 years [6]. The antioxidant activity of clove, capsicum and kalonji seeds is because of eugenol, capsaicinoids and thymol, thymoquinone, carvacrol respectively [7-9]. Moreover, the activity of the antioxidant in the extract is affected by the presence of other components in the extract and there is a scope to improve the antioxidant activity of the extract using synergistic effect with combination of these extracts [10].

Besides possessing the antioxidant activity, these natural extracts possess medicinal benefits as well. Clove has been reported to possess therapeutic use in dental application [11]. It has antiprotozoal antibacterial and anti-inflammatory activity [12-14]. It is known to possess immunomodulatory effects [15]. Equally, capsicum has been valued for its various medicinal actions including antithrombotic, anti-inflammatory and analgesic effects. It has cardio protective, thermogenic influence and also beneficial effects on gastrointestinal system. Capsicum has the potential for pain relief, cancer prevention and weight loss [16]. Likewise, the seeds of kalonji have various curative actions. Haseena et al. [17] reviewed the positive effect of kalonji seeds on male reproductive system. Recently, Randhawa, & Alenazi have also listed analgesic, antipyretic, anti-parkinsonism, anticonvulsant, antidepressant, anti-anxiety, neuroprotective effects

of kalonji seeds [18]. They have also elaborated the effect of the same on learning, memory and encephalitis. Thus, by using these natural ingredients as a source of antioxidant, their therapeutic benefits can be accomplished simultaneously. It was suggested previously that better results are often achieved by using a synergistic activity of antioxidants [19].

Oleoresin, obtained by extraction from plant, is a mixture of essential oil and resin. It is a natural material, which contains different constituents. The present work deals to envisage the antioxidant activity of clove oleoresin and effect of purification on the antioxidant activity of clove oleoresin. The work is extrapolated to study the synergistic effect of clove oleoresin with kalonji seeds extract and capsicum oleoresin.

MATERIALS AND METHODS

Materials

Refined Sunflower Oil (RSFO) was received as a gift sample from M/s Cargill India Pvt. Ltd., New Delhi. Capsicum oleoresin, pure capsaicin and clove oleoresin were obtained from Kancor India Ltd., Angamaly South, India. Kalonji seeds were procured from local vendor. All other chemical reagents and solvents were obtained from s.d. fine-CHEM LiMiTEd, Mumbai.

Analysis of capsicum oleoresin and clove oleoresin

The presence of capsaicinoids was checked by UV spectrometry (Shimadzu UV-1601) of capsicum oleoresin in isopropanol [20]. The capsaicinoids content in capsicum oleoresin was determined by using capsaicin as a standard. The presence of eugenol in clove oleoresin was detected by HPTLC using mobile phase, n-hexane: ethyl acetate (3:1 v / v) [21]. The spots were visualized in UV chamber at 254 nm.

Extraction of kalonji seeds

Kalonji seeds (50 g) were crushed and extracted using ethanol in Soxhlet extractor (400 mL capacity) for 3 hours. The crude Kalonji seeds Ethanol Extract (KEE) was obtained after removal of solvent using rotary vacuum evaporator.

Fractionation of clove oleoresin to Eugenol Rich Fraction (ERF)

Previously dried silica gel (60-120 mesh) weighing 180 g was slurried in n-hexane and filled in a glass column of 3.3 cm internal diameter and 35 cm height. Clove oleoresin weighing 9.0 g was dissolved in 10 mL of n-hexane and loaded on the silica gel column. Further the column was continuously eluted with n-hexane: ethyl acetate (3:1 v / v) at 1 mL / min. Each 10 mL fraction was separately



collected and analyzed by HPTLC on silica gel plates (Merck DC Kieselgel 60 F₂₅₄) using n-hexane: ethyl acetate (3:1 v / v) as mobile phase [21]. The spots were visualized in UV chamber at 254 nm. The fractions rich in eugenol were combined and solvent was removed to get ERF. The concentration of eugenol in ERF was determined by Gas Chromatography at Sophisticated Analytical Instrument Facility (SAIF), IIT-Bombay and Mumbai, India.

Study of antioxidant activity

The various blends of oil containing combinations of capsicum oleoresin (2.5% w / v), clove oleoresin (0.5%, 1.5% w / v) and KEE (2% w / v) were prepared. (% w/v implies g oleoresin in 100 mL RSFO). The oxidative stability of oil blends was checked at 60°C for 30 days at regular interval of 5 days according to the AOCS Official Methods by peroxide value (in meq/kg) (PV, Method Cd 8-53), *p* - anisidine value (*p* - A.V., Method Cd 18 - 90) and Total Oxidation (Totox) value (in meq / kg) (Method Cg 3-91) [22]. All the experiments were carried out in triplicate and the values were expressed as arithmetic mean of the experiments along with standard deviation. At 60°C, the rate of oxidation is raised but the mechanism of oxidation remains identical as that of the ambient temperature [23].

The antioxidant activities of natural antioxidants at above mentioned concentrations were compared with the control sample and TBHQ (200 ppm or mg/L) under the same conditions. The relative antioxidant activities were compared using Oxidative Factor (OXF) for antioxidants using following formula,

$$OXF = \frac{(PV_{final} - PV_{initial})_{antioxidant}}{(PV_{final} - PV_{initial})_{control}}$$

Where, PVs indicate the mean values of all triplicate determinations of the peroxide value [24].

RESULTS AND DISCUSSION

Analysis of capsicum oleoresin and clove oleoresin

The presence of capsaicinoids in capsicum oleoresin was confirmed by the peak at 281 nm in the UV spectrum given by the solution of capsicum oleoresin in isopropanol [20]. The capsaicinoids content in capsicum oleoresin was found to be 3.14%. Clove oleoresin contains eugenol as an active constituent along with other constituents (Figure 1). The eugenol content in clove oleoresin was detected as 66.7% by HPTLC, calculated by peak area. The Rf value of the eugenol was obtained to be 0.89. The eugenol was purified by Column Chromatography and analyzed by HPTLC (Figure 2). The purity of eugenol in ERF was found to be 68% by Gas Chromatography.

Activity of clove oleoresin and ERF in RSFO

Table 1, table 2 and figure 3 indicate the effect of clove oleoresin and ERF on oxidative stability of RSFO. Clove oleoresin showed pro-oxidant activity whereas ERF showed antioxidant activity.

The pro-oxidant effect of clove is already noticed [25,26]. Eugenol generates the reactive oxygen species under alkaline conditions, indicating under alkaline conditions conditions, eugenol acts as a pro-oxidant [27]. The oxidation of eugenol to vanillin under basic conditions is a developed process, where isomerization of eugenol to isoeugenol takes place initially [28]. It can be predicted from the structures of isoeugenol and vanillin that the vanillin formation may take place via peroxide linkage generation at the allylic position of isoeugenol similar to curcumin [29] (Figure 4). Thus, it implies that

non-basic or polar environment has to be employed to stabilize the eugenol. The stabilization of eugenol and unveiling its antioxidant behaviour are correlated. The non-polar constituents were detected in clove oleoresin (Figure 1). On the other hand, oil used in the study was refined with acid value 0.28 mg KOH / g. Hence, it lacked the

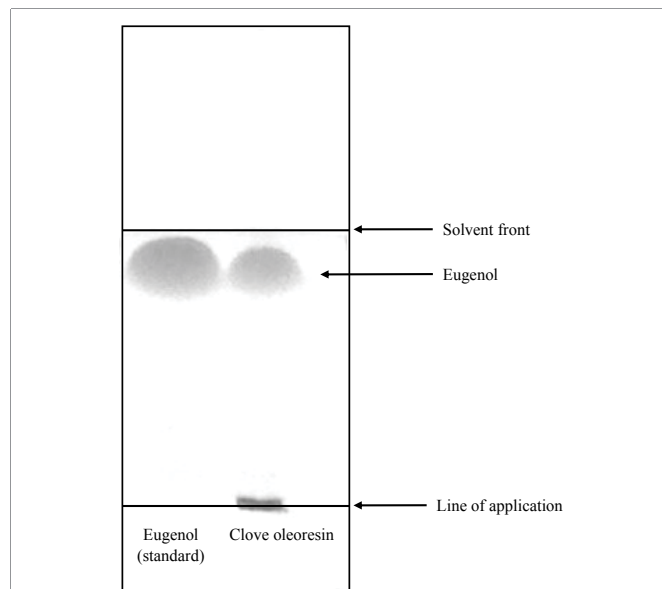


Figure 1: Thin layer chromatography of standard eugenol and clove oleoresin.

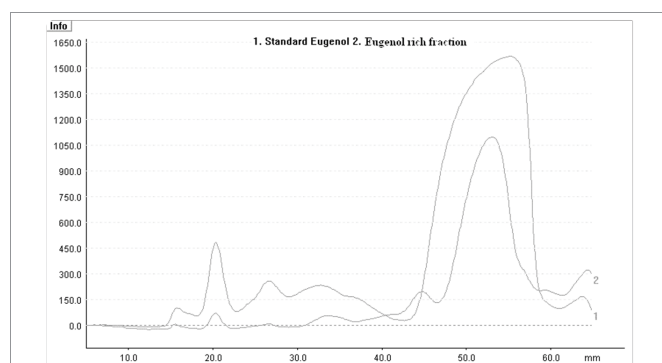


Figure 2: Overlay of HPTLC chromatogram of standard eugenol and eugenol rich fraction.

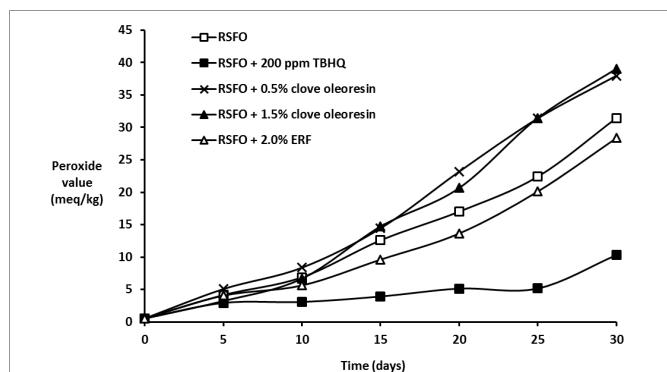
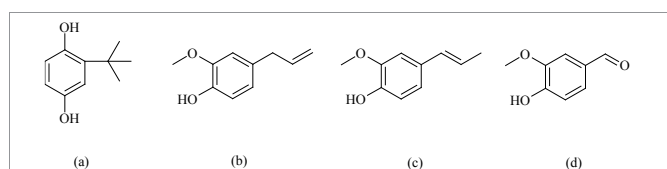
Table 1: Effect of synergism of clove oleoresin with capsicum oleoresin and KEE on oxidative stability (peroxide formation) of refined sunflower oil without antioxidant (RSFO) at 60°C.

Sample	Time (days)	OXF					
		5	10	15	20	25	30
RSFO + 200 ppm TBHQ		0.66	0.40	0.28	0.28	0.21	0.32
RSFO + 0.5% clove oleoresin		1.26	1.23	1.15	1.37	1.41	1.21
RSFO + 1.5% clove oleoresin		0.74	0.97	1.18	1.22	1.41	1.25
RSFO + 1% ERF		0.93	0.82	0.77	0.80	0.96	0.98
RSFO + 2% ERF		0.97	0.80	0.75	0.80	0.90	0.90
RSFO + 1.5% clove oleoresin + 2.5% capsicum oleoresin		0.74	0.93	0.92	0.88	0.81	0.80
RSFO + 2% KEE + 1.5% clove oleoresin		0.43	0.48	0.54	0.57	0.51	0.53
RSFO + 2% KEE + 2.5% capsicum oleoresin		0.48	0.45	0.43	0.54	0.47	0.53


Table 2: Effect of synergism of clove oleoresin with capsicum oleoresin and KEE on oxidative stability (totox value) of refined sunflower oil without antioxidant (RSFO) at 60°C.

Sample	Time (days)	Totox [†]					
		5	10	15	20	25	30
RSFO		17.0 ± 1.4	22.6 ± 1.5	30.0 ± 1.5	40.1 ± 1.6	51.5 ± 2.2	70.9 ± 3.9
RSFO + 200 ppm TBHQ		6.3 ± 0.9	8.2 ± 0.8	11.2 ± 1.0	14.0 ± 1.1	14.5 ± 1.6	26.3 ± 2.8
RSFO + 0.5% clove oleoresin		20.9 ± 0.7	24.8 ± 0.8	30.6 ± 1.7	43.1 ± 0.9	64.6 ± 3.2	80.5 ± 3.5
RSFO + 1.5% clove oleoresin		20.8 ± 0.8	24.7 ± 1.1	30.4 ± 0.9	42.5 ± 2.1	64.3 ± 2.2	79.7 ± 3.3
RSFO + 1% ERF		19.2 ± 1.2	22.7 ± 0.9	27.9 ± 1.1	41.6 ± 1.7	57.7 ± 2.5	67.7 ± 5.3
RSFO + 2% ERF		18.1 ± 0.7	21.5 ± 0.5	26.4 ± 1.4	38.3 ± 0.4	52.0 ± 1.1	61.5 ± 3.6
RSFO + 1.5% clove oleoresin + 2.5% capsicum oleoresin		8.3 ± 0.2	9.6 ± 0.3	11.6 ± 0.4	15.1 ± 0.9	18.2 ± 0.5	25.2 ± 0.5
RSFO + 2% KEE + 1.5% clove oleoresin		5.2 ± 0.3	6.0 ± 0.1	7.1 ± 0.4	9.9 ± 0.2	11.7 ± 0.3	16.8 ± 0.6
RSFO + 2% KEE + 2.5% capsicum oleoresin		4.3 ± 0.1	4.9 ± 0.2	5.7 ± 0.1	9.4 ± 0.4	10.9 ± 0.4	17.0 ± 0.7

[†] The values given are means of three consecutive experiments ± standard deviations


Figure 3: Effect of clove oleoresin on oxidative stability (peroxide formation) of RSFO at 60°C.

Figure 4: Chemical structures of (a) TBHQ (b) Eugenol (c) Isoeugenol (d) Vanillin.

polar constituents, i.e. free fatty acids, which gets removed during refining process. Subsequently, scarcity of polar medium facilitated the degradation of eugenol. Therefore, OXF and Totox values of blends containing clove oleoresin were higher than RSFO.

During purification of clove oleoresin to isolate ERF, non-polar constituents were retained on the column. Thus, eugenol gets stabilized because of the absence of non-polar constituents. Hence, ERF revealed antioxidant activity. However, the results proved that the antioxidant activity of ERF was much lower than TBHQ even at higher concentration. This can be explained on the basis of structural difference between TBHQ and eugenol (Figure 4). TBHQ and eugenol are phenolic hydroxyl compounds. The phenolic compounds with electron donating group at ortho or para position are better antioxidants than those with meta-electron donating group [30,31]. Also, the size of the electron donating group like alkyl group

mainly affects the antioxidant potential of the phenolic compounds [32]. TBHQ contains a bulky tertiary butyl group at ortho position to phenolic hydroxyl group and a hydroxyl group at para position to phenolic hydroxyl group. Thus, it contains two strong activating groups at ortho and para positions to phenolic hydroxyl group. In comparison to TBHQ, eugenol contains a moderately activating methoxy group at ortho position to phenolic hydroxyl group and a weakly activating allyl group at para position to phenolic hydroxyl group. Hence, the radical scavenging activity of eugenol is weak as compared to TBHQ. This implies that position as well as the electron donating strength of electron donating group is responsible for the antioxidant potential of a phenolic hydroxyl compound.

Synergistic antioxidant activity of clove oleoresin, capsicum oleoresin with KEE

Table 1, table 2 and figure 5 represent the synergistic antioxidant activity of clove oleoresin with capsicum oleoresin and KEE in RSFO. They also signify the synergistic antioxidant activity of capsicum oleoresin with KEE in RSFO.

Capsicum oleoresin (2.5% w/v) displayed synergistic antioxidant activity with lecithin and curcuminoids in RSFO [33]. Similarly, KEE (2% w/v) exhibited synergistic antioxidant activity with curcuminoids in RSFO [34]. The combined effect of clove oleoresin with capsicum oleoresin illustrated moderate antioxidant activity whereas that of clove oleoresin with KEE revealed remarkable antioxidant activity in RSFO.

Capsicum is reported to contain phenolic acids [35]. Nevertheless, the amount was incapable to overwhelm the impact of non-polar constituents present in clove oleoresin. Since polar medium was not provided, it must be possible that the antioxidant potential of capsicum oleoresin was utilized by both, RSFO and clove oleoresin to achieve stability. Though the OXF of the blend of clove oleoresin with capsicum oleoresin confirmed that the antioxidant activity was not much higher, the Totox values of the blend were almost identical as those of TBHQ. This implies that the blend has the ability to stabilize peroxy compounds thereby inhibiting the formation of secondary degradation products like carbonyl compounds.

In the blend of clove oleoresin with KEE, the polar medium required to achieve the stability of eugenol was provided by KEE due to the presence of free fatty acids in it [9]. The oxidative stability of the blend depends on the concentration of polar and non-polar

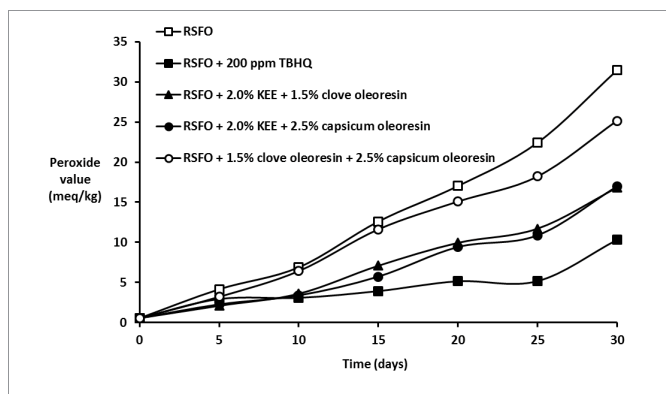


Figure 5: Effect of clove oleoresin in synergism with KEE and capsicum oleoresin on oxidative stability (peroxide formation) of RSFO at 60°C.

constituents. The concentration of polar constituents was increased by means of KEE. The polar constituents present in KEE surpassed the influence of non-polar constituents present in clove oleoresin. Consequently, eugenol present in clove oleoresin did not undergo degradation expressing its antioxidant activity. Hence, the synergistic antioxidant activity of the combination was notable.

Unlike clove oleoresin, capsicum oleoresin displayed antioxidant activity in RSFO, irrespective of the addition of KEE [33]. However, the combined effect of KEE and capsicum oleoresin indicated the similar antioxidant activity as that of KEE and clove oleoresin. The other constituents present in KEE besides antioxidants remained dormant when it is used in combination with capsicum oleoresin. The importance of these constituents is highly focused when they are utilized to generate a specific medium as required for clove oleoresin. Evidently, KEE acts as a good synergistic agent not only by imparting its own antioxidant activity but also by providing the polar medium to stabilize an antioxidant to avail its antioxidant activity.

Both the blends, i.e., clove oleoresin with KEE and capsicum oleoresin with KEE showed low OXF, indicating that these two blends have the ability to prevent peroxide formation in RSFO. Since peroxide formation is low, generation of secondary oxidation products gets automatically reduced. Ultimately, the Totox values of the blends being low, restore the nutritive value of EFAs in RSFO.

CONCLUSIONS

The study of antioxidant activity of natural extracts is often unpredictable due to the presence of other constituents present in the extract. It is quite possible that a constituent possessing an antioxidant activity is present in the extract. However, because of the presence of other constituents, the extract as a whole exhibits pro-oxidant effect. This fact reveals that the antioxidant activity of an extract can be effectively utilized by suppressing the role of other constituents, by means of other extract. Thus, KEE inhibits the effect of non-polar constituents present in clove oleoresin. Hence, the antioxidant activity of the main constituent present as an antioxidant in clove oleoresin, i.e., eugenol is accomplished using KEE.

Moreover, the non-toxicity of these natural antioxidants is valuable for various edible products including vegetable oils. The activity of natural antioxidants as such was not comparable to synthetic antioxidants due to structural variation. However, two or more natural antioxidants can be used in proper synergism for better activity. KEE showed good synergistic antioxidant activity with capsicum oleoresin. Capsicum oleoresin can be used to suppress

the degradation of eugenol. Consequently, the oxidative stability of the sunflower oil can be interpreted as the combined effect of all the constituents and the synergistic activity of the natural extracts can be used. The proper combination and proportion of natural antioxidants can win over the effect of other constituent present in the extract. Accordingly, the benefits of the natural extract can be utilized without purification.

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