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Research Article

Effects of Mutations of Phosphorylation Sites in Sars-Cov-2 Encoded Nucleocapsid Protein on its Dimerization -

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ABSTRACT

Mutations in several phosphorylation sites within the phosphorylation rich domain of SARS-COV-2 Nucleocapsid protein (NCp), including phospho-serines 186, 197 and 202, and C-TAK1 phosphorylation recognition sites phospho-serine 197 within the motif RNpSTP, and arginine 203 and glycine 204 within the motif RGTpSP have been described and have been proposed to prevent the binding and sequestration of NCp by Protein 14-3-3. Here, through structure modeling and thermodynamic calculation, it is shown that mutations of serines 186, 197 and 202 to phenylalanine, leucine and asparagine, and arginine/glycine 203/204 to lysine/arginine or lysine/threonine resulted in significant stabilization of the NCp-NCp complex through increase in Stability Energy ($\Delta G_{\text{stability energy}}$) and Binding Energy ($\Delta\Delta G_{\text{binding energy}}$). These results suggest that SARS-COV-2 has evolved through the above described mutations in NCp to enhance NCp's dimerization and its activity as an essential co-factor for the replication, transcription and packaging of the SARS-COV-2 genome.

INTRODUCTION

SARS-COV-2 Nucleocapsid protein (NCp) plays an essential role as a co-factor in the initiation and control of the replication, transcription and packaging of the SARS-COV-2 genome [1-6]. Dimerization of SARS-COV-2 Nucleocapsid protein (NCp) is an important step in the replication, transcription and packaging of the SARS-COV-2 genome [7-12]. Oligomerization of NCp is also required for the packaging of SARS-COV-2 genome [8-12]. Phosphorylation of NCp within a phosphorylation rich domain has been proposed to play an important role in the control of NCp functions [3,13-18]. Recent work has described the existence of a cellular response mechanism for preventing dimerization of NCp involving phosphorylation dependent sequestration of monomeric NCp by Protein 14-3-3 [13,18]. The phosphorylation rich domain located in the linker region of NCp is exposed at its surface and contains two recognition and binding sites (RNpSTP and RGTpSP) for Protein 14-3-3 [13,18]. Other phosphorylation sites, including serines 186 and 202, and threonines 198 and 205 are also present in the phosphorylation rich domain of NCp [13,18].

Phosphorylation sites 186, 197 and 202 are mutated to phenylalanine, leucine and asparagine in SARS-COV-2 strains/substrains from Iran, Spain and India respectively [13]. Phosphorylation recognition sites within the RNpSTP and RGTpSP motifs that are recognized by C-TAK1 [13,18-21], including phospho-serine 197, arginine 203 and glycine 204 are mutated to leucine, lysine and arginine or lysine and threonine in SARS-COV-2 strains/substrains isolated from Israel, Italy, Poland, Bangladesh, Greece and Czech republic [13]. The significance of these mutations has not been clearly established. Here, we use structure model analysis, and thermodynamic calculation to study the effects of these mutations on the dimerization of NCp. The results provide evidence that mutations in the phosphorylation sites and phosphorylation recognition sites phospho-serines 186, 197 and 202, arginine 203 and glycine 204 increase the Stability Energy ($\Delta G_{\text{stability energy}}$) and Binding Energy ($\Delta\Delta G_{\text{binding energy}}$) of NCp-NCp complex. It is submitted that SARS-COV-2 has evolved to optimize the dimerization of NCp through mutations of at least 3 phosphorylation sites, phospho-serines 186, 197 and 202 and phosphorylation recognition sites, phospho-serine 197, arginine 203 and glycine 204.

METHODS

The de novo rendering of the structure of dephospho-NCp using the Quark Program pursuant to Xu and Zhang, et al. [22,23] was as described previously [13]. Phosphorylation of SARS-COV-2 Nucleocapsid protein (NCp) was performed with FoldX using the Build Model Program pursuant to Guerois et al. and Schymkowitz, et al. [24,25]. Mutation of amino acid residues within NCp was performed with FoldX using the Build Model program pursuant

Guerois, et al. and Schymkowitz, et al. [24,25]. Docking experiments to identify the dimerization of NCp were performed using the ZDOCK program pursuant to Pierce, et al. [26]. NCps rendered in this work and Protein 14-3-3 (1YZ5) based on the structure determination of Benzinger, et al. [27] were analyzed and visualized by the CCP4 Molecular Graphics Program Version 2.10.11 as described by Mc Nicolas, et al [28] and the ZMM Molecular Modeling Program as described by Garden and Zhorov [29].

Determination and calculation of Stability Energy ($\Delta G_{\text{stability energy}}$) of the protein complexes was performed using the Stability Program of FoldX as described by Guerois, et al. and Schymkowitz, et al. [24,25]. Binding Energy ($\Delta\Delta G_{\text{binding energy}}$) of protein complex was determined and calculated using the Analyze Complex Program of FoldX as described by Guerois, et al. and Schymkowitz, et al. [24,25]. Binding Energy Difference ($\Delta\Delta\Delta G_{\text{binding energy difference}}$) between phospho-NCps complex and mutant phospho-NCps complex was calculated pursuant to Teng, et al. and Nishi, et al. [30-32] from the equation: $\Delta\Delta\Delta G_{\text{binding energy difference}} = \Delta\Delta G_{\text{binding energy}}$ (binding energy of phospho-NCp-14-3-3 complex) - $\Delta\Delta G_{\text{binding energy}}$ (binding energy of dephospho-NCp-14-3-3 complex).

RESULTS

The effects of mutations in the phosphorylation sites of NCp were determined with respect to the binding of dephospho-NCp, mutant dephospho-NCp, phospho-NCp and mutant phospho-NCp to themselves. Figure 1 and figure 2 represent docking experiments of dephospho-NCp and mutant S186F-NCp with themselves. Both dephospho-NCp and mutant S186F-NCp readily form complexes. Major difference in conformations of dephospho-NCp and mutant S186F-NCp could be observed. The Stability Energy ($\Delta G_{\text{stability energy}}$) and Binding Energy ($\Delta\Delta G_{\text{binding energy}}$) of dephospho-NCp dimer were calculated to be ~698 Kcal/mol and 91 Kcal/mol respectively while the calculated Stability Energy ($\Delta G_{\text{stability energy}}$) and Binding Energy ($\Delta\Delta G_{\text{binding energy}}$) of mutant S186F-NCp dimer was calculated to be ~770 Kcal/mol and ~168 respectively. These results suggest that mutation of serine 186 of NCp to phenylalanine 186 was accompanied by enhanced stabilization and binding affinity between the two components of the NCp dimer. The Binding Energy Difference ($\Delta\Delta\Delta G_{\text{binding energy difference}}$) between dephospho-NCp dimer and mutant S186F-NCp dimer was calculated to be ~78 Kcal/mol. Pursuant to Teng et al. and Nishi et al. [30-32], positive Binding Energy Difference ($\Delta\Delta\Delta G_{\text{binding energy difference}}$) is thermodynamic evidence of stabilization and increase of binding efficiency.

Table 1 depicts the thermodynamic calculations of Stability Energy ($\Delta G_{\text{stability energy}}$), Binding Energy ($\Delta\Delta G_{\text{binding energy}}$) and Binding Energy Difference ($\Delta\Delta\Delta G_{\text{binding energy difference}}$) of the dimerization of various NCp mutants with respect to dephospho-NCp dimer. The dimers of mutants S197L and S202N were associated with decreased Stability

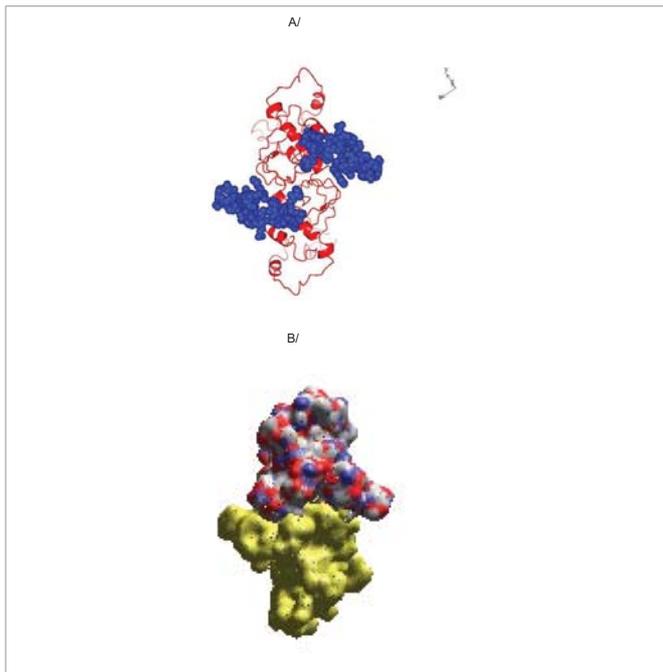


Figure 1: A: Ribbon structure (red) of dephospho-SARS-COV-2 Nucleocapsid protein (NCp) dimer, rendered as described in Method Section. Phosphorylation rich domain is shown as blue spheres. B: Realistic rendering of dephospho-SARS-COV-2 Nucleocapsid protein (NCp) complex (component 1 is colored red and blue while component 2 is colored yellow).

Table 1: Thermodynamic calculations, including Stability Energy ($\Delta G_{\text{stability}}$), Binding Energy ($\Delta\Delta G_{\text{binding}}$) and Binding Energy Difference ($\Delta\Delta\Delta G_{\text{binding difference}}$) that underlie the dimerization of dephospho-NCp and various mutant-NCps.

	Stability Energy (ΔG) Kcal/mol	Binding Energy ($\Delta\Delta G$) Kcal/mol	Binding Energy Difference ($\Delta\Delta\Delta G$) Kcal/mol
-Dephospho-NCp-NCp complex	~698	~90	~ 000
-S186F mutant-NCp-NCp complex	~770	~168	~ 078
-Phospho-serine 186-S197L mutant-NCp-NCp complex	~689	~87	~ -03
-Phospho-serine 186-S202N mutant-NCp-NCp complex	~675	~67	~ -23
-Phospho-serine 186-RG203/204KR mutant-NCp-NCp complex	~737	~136	~ 046
-Phospho-serine 186-RG203/204KT mutant-NCp-NCp complex	~759	~156	~ 066

Energy ($\Delta G_{\text{stability}}$) and Binding Energy ($\Delta\Delta G_{\text{binding}}$). However, the dimers of mutants RG203/204KR-NCp and RG203/204KT-NCp were associated with significant increase of Stability Energy ($\Delta G_{\text{stability}}$), Binding Energy and positive Binding Energy Difference ($\Delta\Delta\Delta G_{\text{binding difference}}$) in comparison to dephospho-NCp dimer. These results are indicative of enhanced binding affinity between the NCp mutants.

Figure 3 illustrates the docking of phospho-serine 186-NCp. There was considerable difference in the conformations of phospho-serine 186-NCp and mutant S186F-NCp dimers (compare figure 2 with

figure 3). The calculated Stability Energy ($\Delta G_{\text{stability}}$) and Binding Energy ($\Delta\Delta G_{\text{binding}}$) of phospho-serine 186 NCp dimer were ~667 Kcal/mol and 63 Kcal/mol respectively while the calculated Stability Energy ($\Delta G_{\text{stability}}$) and Binding Energy ($\Delta\Delta G_{\text{binding}}$) of mutant S186F-NCp dimer was ~770 Kcal/mol and ~168 respectively. These results are evidence that mutation of phospho-serine 186 of NCp to phenylalanine 186 was accompanied by enhanced stabilization and binding affinity between the two components of the NCp dimer. The Binding Energy Difference ($\Delta\Delta\Delta G_{\text{binding difference}}$) between phospho-serine 186-NCp dimer and mutant S186F-NCp dimer was calculated to be ~105 Kcal/mol, confirming thermodynamic stabilization and enhancement of binding efficiency between the molecules of the NCp dimer.

Figures 4 and 5 show the dimerizations of phospho-serine 197-NCp and phospho-serine 197-S186F mutant-NCp. The results show that both phospho-serine 197-NCp and phospho-serine 197-S186F mutant-NCp form dimers and there was marked difference in their conformations. The calculated Stability Energy ($\Delta G_{\text{stability}}$) and Binding Energy ($\Delta\Delta G_{\text{binding}}$) of phospho-serine197-NCp dimer were ~699 Kcal/mol and ~87 Kcal/mol respectively whereas the Stability Energy ($\Delta G_{\text{stability}}$) and Binding Energy ($\Delta\Delta G_{\text{binding}}$) of the phospho-serine197-S186F mutant-NCp dimer were ~788 Kcal/mol and ~184 Kcal/mol respectively. The Binding Energy Difference ($\Delta\Delta\Delta G_{\text{binding difference}}$) between phospho-serine 197-NCp dimer and phospho-serine197-S186F mutant-NCp dimer was calculated to be ~97 Kcal/mol, evidencing enhanced thermodynamic stabilization and binding affinity between the components of the phospho-serine197-S186F mutant-NCp in the NCp dimer.

Figures 6 and 7 summarizes the docking experiments of phospho-serine 202-NCp and phospho-serine 202-S186F mutant-NCp with themselves. Both phospho-serine 202-NCp and phospho-serine 202-S197L mutant-NCp form dimers with recognizable difference

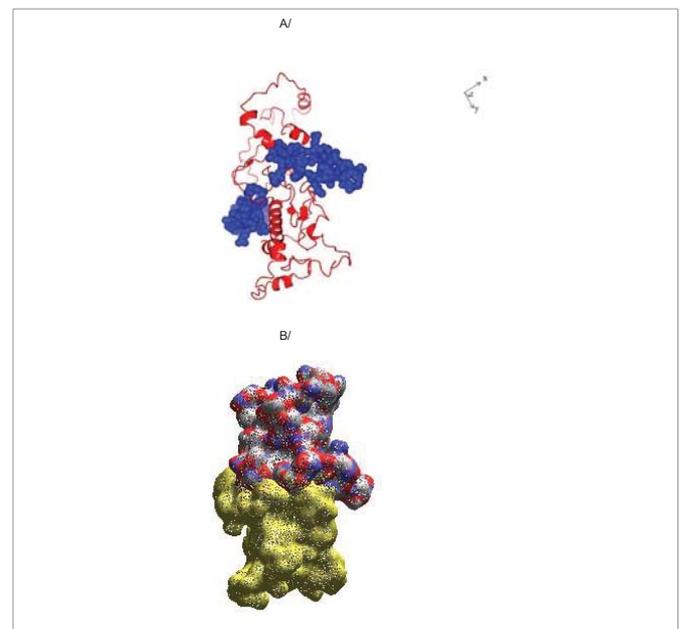


Figure 2: A: Ribbon structure (red) of S186F mutant-SARS-COV-2 Nucleocapsid protein (NCp) and Protein 14-3-3 complex, rendered as described in Method Section. Blue spheres show the phosphorylation rich domain. B: Realistic rendering of S186F mutant SARS-COV-2 Nucleocapsid protein (NCp) complex (component 1 is colored red and blue while component 2 is colored yellow).

in the conformation of their respective dimers. The Stability Energy ($\Delta G_{\text{stability energy}}$) and the Binding Energy ($\Delta\Delta G_{\text{binding energy}}$) of the phospho-serine 202-NCp dimer were calculated to be ~ 696 Kcal/mol and 95 Kcal/mol respectively. The Stability Energy ($\Delta G_{\text{stability energy}}$) and the Binding Energy ($\Delta\Delta G_{\text{binding energy}}$) of the phospho-serine 202-S186F mutant-NCp dimer were calculated to be 761 Kcal/mol and ~ 165 Kcal/mol respectively. The Binding Energy Difference ($\Delta\Delta\Delta G_{\text{binding}}$

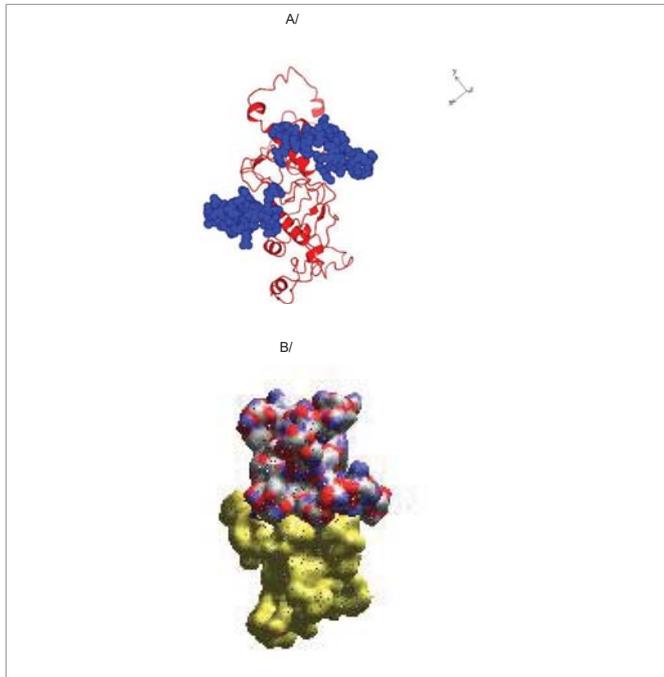


Figure 3: A: Ribbon structure (red) of phospho-serine 197-SARS-COV-2 Nucleocapsid protein (NCp) dimer. Blue spheres show the phosphorylation rich domain of NCp), rendered as described in Method Section. B: Realistic rendering of phospho-197-SARS-COV-2 Nucleocapsid protein (NCp) dimer (component 1 is colored red and blue while component 2 is colored yellow).

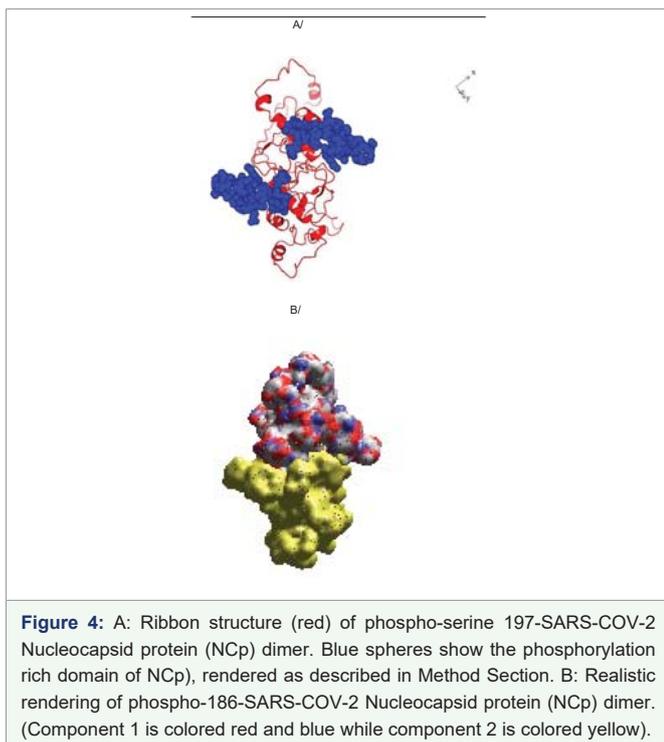


Figure 4: A: Ribbon structure (red) of phospho-serine 197-SARS-COV-2 Nucleocapsid protein (NCp) dimer. Blue spheres show the phosphorylation rich domain of NCp), rendered as described in Method Section. B: Realistic rendering of phospho-186-SARS-COV-2 Nucleocapsid protein (NCp) dimer. (Component 1 is colored red and blue while component 2 is colored yellow).

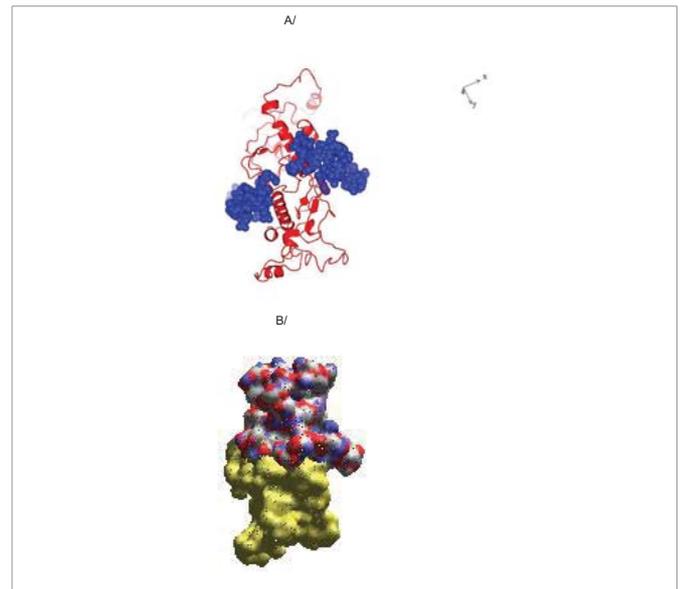


Figure 5: A: Ribbon structure (red) of phospho-serine 197-S186F mutant-SARS-COV-2 Nucleocapsid protein (NCp) dimer. Blue spheres show the phosphorylation rich domain of NCp), rendered as described in Method Section. B: Realistic rendering of phospho-186-SARS-COV-2 Nucleocapsid protein (NCp) dimer. (Component 1 is colored red and blue while component 2 is colored yellow).

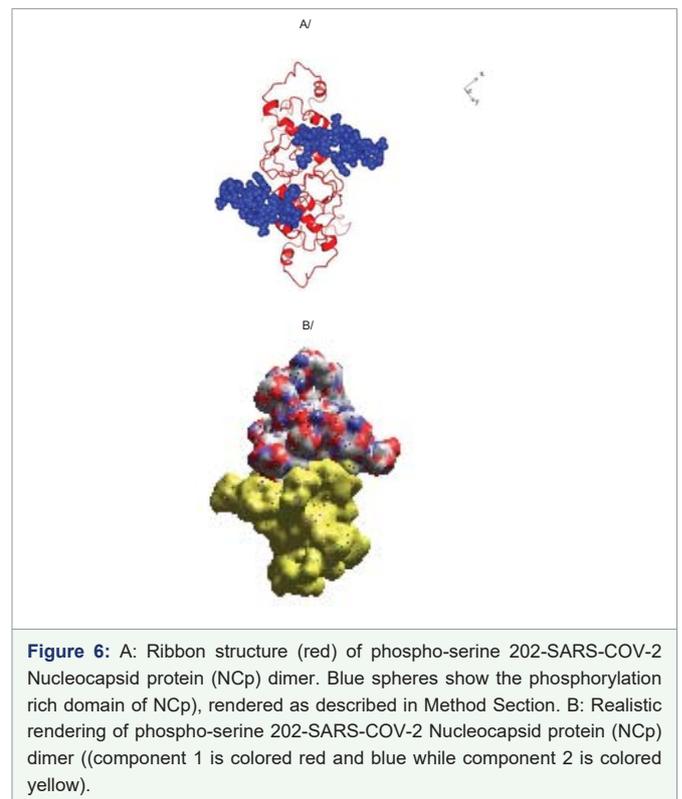


Figure 6: A: Ribbon structure (red) of phospho-serine 202-SARS-COV-2 Nucleocapsid protein (NCp) dimer. Blue spheres show the phosphorylation rich domain of NCp), rendered as described in Method Section. B: Realistic rendering of phospho-serine 202-SARS-COV-2 Nucleocapsid protein (NCp) dimer ((component 1 is colored red and blue while component 2 is colored yellow).

energy difference) between the dimers of phospho-serine 202-NCp and phospho-serine 202-S197L mutant-NCp was ~ 70 Kcal/mol. These results are consistent with the conclusion that mutation of serine 186 to phenylalanine caused enhanced stability and binding affinity of between the components in the NCp dimer.

Table 2 summarizes the thermodynamic calculations of various

NCp mutant-Protein 14-3-3 complexes. The results showed that like mutations of serines 186, 197 and 202 of NCp, mutations of arginine 203 and glycine 204 within the motif RGTpSP to lysine 203 and arginine 204 or lysine 203 and threonine 204 were accompanied by significant increase in Stability Energy ($\Delta G_{\text{stability energy}}$) and Binding Energy ($\Delta\Delta G_{\text{binding energy}}$). Calculation of Binding Energy Difference ($\Delta\Delta\Delta G_{\text{binding energy difference}}$) between the various reference phospho-NCps and mutant-NCps confirms that the above mutations were indeed accompanied by significant increase in stability and binding efficiency of the respective NCps.

DISCUSSION

It was previously shown that phosphorylation of serines 186, 197 and 202 of NCp cause a decrease in Stability Energy and Binding Energy of NCp in its dimer [13,18]. Mutations in the phosphorylation sites, serine 186, 197 and 202, and phosphorylation recognition sites, phospho-serine 197, arginine 203 and glycine 204 within the motifs RNpSTP and RGTpSP to leucine, lysine 203 and arginine 204 or lysine 203 and threonine 204 respectively were identified in strains/sub-strains isolated from individuals located in various parts of the world [13]. It was proposed that mutations in the phosphorylation sites and phosphorylation recognition motifs allow NCp to evade sequestration by Protein 14-3-3 which would result in enhanced dimerization of NCp, an important essential step for NCp to act as a co-factor for the replication, transcription and packaging of the SARS-COV-2 genome [13,18]. The present work provides evidence that mutations in the phosphorylation sites, serines 186, 197 and 202, and phosphorylation recognition sites, phospho-serine 197, arginine 202 and glycine 203 within the motifs RNpSTP and RGTpSP were accompanied by significant enhancement of the stability and binding affinity of the NCp dimer.

While there is a phosphorylation dependent cellular response mechanism to bind, sequester and inhibit the functions of NCp by

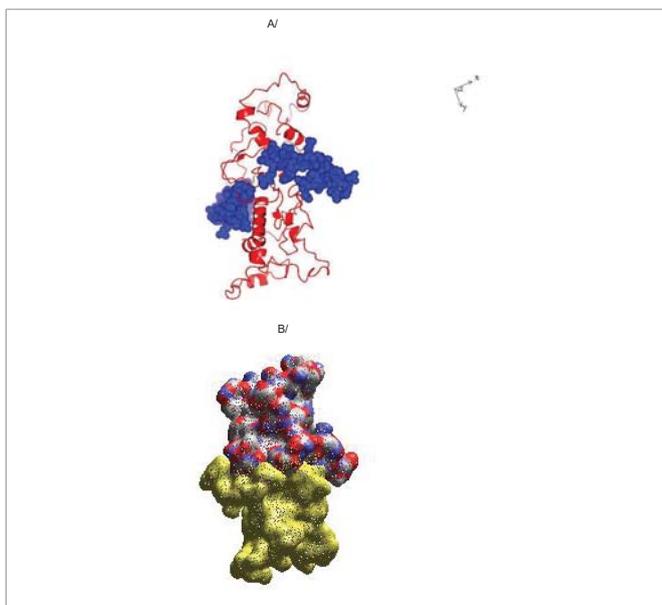


Figure 7: A: Ribbon structure (red) of phospho-serine 202-S186F mutant-SARS-COV-2 Nucleocapsid protein (NCp) dimer. Blue spheres show the phosphorylation rich domain of NCp, rendered as described in Method Section. B: Realistic rendering of phospho-202-S186F mutant-SARS-COV-2 Nucleocapsid protein (NCp) dimer (component 1 is colored red and blue while component 2 is colored yellow).

Table 2: Thermodynamic calculations, including Stability Energy ($\Delta G_{\text{stability}}$), Binding Energy ($\Delta\Delta G_{\text{binding energy}}$) and Binding Energy Difference ($\Delta\Delta\Delta G_{\text{binding energy difference}}$) that underlie the dimerization of various phospho-NCp and various mutant-NCps.

	Stability Energy (ΔG) Kcal/mol	Binding Energy ($\Delta\Delta G$) Kcal/mol	Binding Energy Difference ($\Delta\Delta\Delta G$) Kcal/mol
-Phospho-serine 186-NCp-NCp complex	~667	~63	~ 000
-S186F mutant-NCp-NCp complex	~770	~168	~ 105
-Phospho-serine 186-S197L mutant-NCp-NCp complex	~660	~62	~ -01
-Phospho-serine 186-S202N mutant-NCp-NCp complex	~658	~55	~ -08
-Phospho-serine 186-RG203/204KR mutant-NCp-NCp complex	~665	~67	~ 004
-Phospho-serine 186-RG203/204KT mutant-NCp-NCp complex	~668	~67	~ 004
-Phospho-serine 197-NCp-NCp complex	~699	~87	~ 000
-S197L mutant-NCp-NCp complex	~689	~87	~ 000
-Phospho-serine 197-S186F mutant-NCp-NCp complex	~788	~184	~ 097
-Phospho-serine 197-S202N mutant-NCp-NCp complex	~844	~239	~ 152
-Phospho-serine 197-RG203/204KR mutant-NCp-NCp complex	~734	~130	~ 043
-Phospho-serine 197-RG203/204KT mutant-NCp-NCp complex	~760	~157	~ 070
Phospho-serine 202-NCp-NCp complex	~696	~95	~ 000
-S202N-mutant-NCp-NCp complex	~689	~87	~ -08
-Phospho-serine 202-S186F mutant-NCp-NCp complex	~761	~165	~ 070
-Phospho-serine 202-S197L mutant-NCp-NCp complex	~737	~141	~ 046
-Phospho-serine 197-RG203/204KR mutant-NCp-NCp complex	~710	~111	~ 016
-Phospho-serine 197-RG203/204KT mutant-NCp-NCp complex	~756	~155	~ 060

Protein 14-3-3 in cells infected with SARS-COV-2, it is submitted that the latter has evolved to evade sequestration by Protein 14-3-3 through mutations of phosphorylation sites, S186F, S197L and S202N, and phosphorylation recognition sites, pS197L, RG203/204KR and RG203/204KT within the phosphorylation motifs, RNpSTP and RGTpSP [13,18]. It is also submitted that SARS-COV-2 has evolved through these mutations to enhance NCp's dimerization and its activity as an essential co-factor for the replication, transcription and packaging of the SARS-COV-2 genome (the present work).

Because of the essential role of NCp dimerization in the initiation and control of the replication, transcription and packaging of the SARS-COV-2 genome, drug discovery and vaccine development



programs that target NCp specifically are quite apparent [33-40]. However, from the above, it is clear that any drug discovery and vaccine development program that target NCp must take into account mutations in the phosphorylation sites (S186F, S197L and S202N) and phosphorylation recognition sites (phospho-serine 197, RG203/204KR and RG203/204KT) that occur within NCp in SARS-COV-2 strains and sub-strains that are seen in various populations and geographical areas. Molecules that act to enhance the sequestration of NCp by Protein 14-3-3 and prevent the dimerization of NCp are potential therapeutics for the control of SARS-COV-2 viability, infection and virulence. Molecules that act to prevent the dimerization of NCp at picomolar concentrations have been identified and are being characterized (Limtung, P. and Tung, H.Y.L., manuscript in preparation) [41,42].

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